task_5ztn80yw2uibcsxr_with_calculation

Student Group

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Table of Contents

conversions, battery, chapter1 1

Exercise E6 Conversions: Battery

1. How many minutes could a consumer with $3~{\rm W}$ get supplied by an ideal battery with $10~{\rm kWh}$?

Result

2. Hard: Why is a real Lithium-ion battery not able to provide the given power for the calculated time?

Solution

$$\begin{align*} W &= 10~{\rm kWh} &&= 10'000~{\rm Wh}\\ \\ t &= {{W }\over{P }} &&= {{10'000~{\rm Wh}}\over{3~{\rm W}}} \\ &&&= 3'333~{\rm h}= 200'000~{\rm min} \quad (= 139~{\rm days}) \end{align*}$$

Result

$$\begin{align*} t = 200'000~{\rm min} \end{align*}$$

There are additional losses:

- The battery has an internal resistance. Depending on the current the battery provides, this leads to internal losses.
- The internal resistance of the battery depends on the state of charge (SoC) of the battery.
- The wires also add additional losses to the system.